S(+)MAGNOFLORINE BROMIDE: ISOLATED FROM CROTON TURUMIOUIRENSIS

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Previous difficulties in isolating pure quaternary alkaloids from Venezuelan C. turumiquirensis (1) prompted us to approach the identification of the bases by X-ray diffraction. After the anion was exchanged from chloride to bromide, a sample of one of the quaternary salts was crystallized from methanol by slow diffusion of acetone. The mother liquors darkened somewhat, but a small quantity of an alkaloid bromide appeared as wellformed prisms.

Monoclinic, prismatic b, C2 (C_2^3) , a = 24.17, b = 8.31, c = 11.60 Å, $\beta =$ 109° 30' . C20H26NO4Br.CH3OH, F.W. 456.4, Z = 4.

The space group was determined from systematic absences by Weissenberg, and precession photographs and intensities were measured on a fully automated Picker four-circle diffracto-Fourier and least-squares meter.



methods afforded structure 1, which is that of magnoflorine bromide (2). A molecule of methanol of crystallization also became apparent during refinement.

The diphenyl system in the aporphine series of alkaloids cannot assume planarity, and, in the case of magnoflorine, the results show a dihedral angle of 29° between the planes of the aromatic rings. The molecule is, of course, optically active; from the positive Cotton effect in the optical rotatory dispersion (3), the Croton base is shown to be S(+) magnoflorine bromide, mp 220° (decomp.), $[\alpha]^{20}D+$ 222° (EtOH). The final value of R is 0.14. Although this is the first alkaloid identified in the species, no further work is planned on the crystal because it is a known compound. Full details of this work are available (4).

Experimental conditions. Source MoK α , $\lambda = 0.7107$ Å, 943 non-zero independent reflections.

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